

# Chemical resistance of SIGRAFLEX® sealing materials



#### Chemical properties

Graphite is insoluble and infusible. It counts as one of the most chemically resistant materials.

#### Organic chemistry

Graphite is resistant to virtually all media in the field of organic chemistry. These typically include, for example, the intermediate and/or final products of the following industries:

- Petrochemistry
- Coal conversion
- Synthetics
- Varnish and paint
- Cosmetics
- Food and stimulants industry
- Photochemicals
- Cooling agents
- Anti-freezing agents

#### Inorganic chemistry

Graphite is resistant to almost all inorganic media as well, for example to many acids and bases, as well as probably all aqueous salt solutions and to most technical gases.

# Medium resistance of sealing material made of SIGRAFLEX flexible graphite foil, stainless steel and pure PTFE

The following media resistance list shall provide an overview. For media which are not included it is generally advised to confer with the sealing manufacturer or with SGL Carbon.

The following table provides information for the materials SIGRAFLEX flexible graphite, stainless steel and pure PTFE. Depending on the composition of the gasket material, either one, or two or all three columns need to be considered.

The resistance data apply to operating temperatures of the medium mentioned which are known to us. However for media being operated at above 450 °C or 842 °F, we generally ask for consultation.

Products containing PTFE must not be used above  $300\,^{\circ}\text{C}$  or  $572\,^{\circ}\text{F}$ .

The information is based on experience and laboratory tests but is partly only analogies. Therefore warranty may not be covered in individual cases. It should be noted, that mixtures can be partly more critical than pure media or vice versa. Four different cases can be distinguished:

- 1. resistant
- 2. not resistant
- 3. limited resistance
- 4. insufficient data -

The third case depends on the stability of operation, operating temperatures or the concentration. It is advised to confer with the sealing manufacturer or with SGL Carbon.

	SIGRAFLEX	Stainless steel 316, 316L, 1.4401,	
Medium	flexible graphite	316, 316L, 1.4401, 1.4404	Pure PTFE
A			
Abietic acid	•		•
Accumulator acid	200°C/392°F: ●		•
Acetal	•	_	•
Acetaldehyde	•	•	•
Acetamide	•		•
Acetanilide (= N-phenylacetamine)	•	_	•
Acetic acid, aqueous solution 50 %	•	100°C/212°F: ●	•
Acetic acid 100 % [= glacial acetic acid]	•	20°C/68°F: ●	•
Acetic acid gas	•		•
Acetic anhydride	•	•	•
Aceto acetic ester	•	•	•
Acetoin (= 3-hydroxid-2-butanone)	•	•	•
Acetone	•	•	•
Acetone cyanohydrin			
[= 2-hydroxy-2-methylpropionitrile = 2-cyanopropane-2-ol]	•	•	•
Acetonitrile	•	•	•
Acetophenone	•	•	•
Acetyl chloride	•	- [p. corr]	•
Acetylene (= ethine = ethyne)	•	•	•
Acetylene tetrachloride	•		•
Acrolein (= properal)	•	•	•
Acrylamide, aqueous solution	•	•	•
Acrylic acid ethyl ester	•	•	•
Acrylic acid, anhydrous	•	•	•
Acrylonitrile	•	•	•
adipic acid	•	•	•
Adipic ester	•	•	•
Adiponitrile	•	•	•
·	400°C/752°F: ●,		
	for higher temperatures please refer to		
	our technical information about tempe-		
Air	rature resistance and ask for advice.	•	•
Alcohol and alcoholic drinks	•	•	•
Alcohols	•	•	•
Aldehyde	•	•	•
Alkylamine	•	•	•
Alkylphenol	•		•

<sup>●</sup> resistant; ■ not resistant; ▲ limited resistance; – insufficient data; p.corr = pitting corrosion

	SIGRAFLEX	Stainless steel 316, 316L, 1.4401,	
Medium	flexible graphite	1.4404	Pure PTF
Alkylsulfonates	•		
Allyl acetate (= propenyl-acetate)	•	_	
Allyl alcohol	•	-	
Allyl bromide	•	_	
Allyl chloride	•	_	
Allyl chloroformate	•	_	
Allyl ether	•		•
Allyl glycidyl ether	•		•
Allyliodide	•		•
Allyl isothiocyanate	•		
Allylamine	•		
Allyltrichlorosilane	•		
Alpha-methylstyrene	•	•	
Alpha-pinene	•	_	
Alum (= aluminium potassium sulfate)	•		
Aluminium alkyl halides			
Aluminium alkyl hydrides			
Aluminium alkyls			
Aluminium chlorate	20 °C / 68 °F: ●	20°C/68°F: ●	
Aluminium chloride	20 07 00 1.0	20 0700 1.	
Aluminium fluoride	•		
Aluminium nitrate			
Aluminium phosphate			
Aluminium sulfate, aqueous solution > 10 %	•		
Aluminium sulfate, aqueous solution 10 %	•	100 °C/212 °F: ●	
Aluminium triethyl		100 07212 1.	
Amidosulfuric acid			
Amine			
Aminophenols			
Ammonia water (= ammonium hydroxide)	-		
Ammonia, liquid		20 C/06 F. •	
Ammonia, gaseous	•		
Ammonium bifluoride, not saturated			
Ammonium bifluoride, not saturated			
<u> </u>			
Ammonium carbonate			
Ammonium chloride (= salmiac)	•	0000/0005	
Ammonium fluoride, anhydrous		20°C/68°F: ●	
Ammonium fluoride, hydrous	•	0000/0005	
Ammonium nitrate		20°C/68°F: ●	
Ammonium nitrite, aqueous solution	•		
Ammonium persulfate, aqueous solution	•		
Ammonium phosphate	•	100°C/212°F: ●	
Ammonium phosphate, aqueous solution			
Ammonium sulfate	•		
Ammonium sulfide, aqueous solution	•		
Ammonium sulfite, aqueous solution > 50 %	•		
Ammonium sulfite, aqueous solution 50 %	•	20°C/68°F: ●	
Amyl acetate	•		
Amyl alcohol	•	<u> </u>	
Amyl mercaptan	•		
Amylamine	•		
Aniline (= aminobenzene)	•		
Anisidine (= methoxyaniline)	•		
Anisole	•		•
Anthranilic acid	•		
Anthraquinone	•		

lacktriangle resistant; lacktriangle not resistant; lacktriangle limited resistance; – insufficient data; p.corr = pitting corrosion

Stainless steel **SIGRAFLEX** 316, 316L, 1.4401, Pure PTFE Medium flexible graphite 1.4404 Antimony butter, aqueous solution [= antimony chloride, aqueous solution] Antimony trioxide • Argon 20°C/68°F: ● Arsenic acid Arsenic chloride Arsenic trioxide • •  $\blacksquare$ Asphalt • • В 20 °C / 68 °F: ● Barium chloride, saturated (p. corr) Barium hydroxide Barium salt, aqueous solution Beer Benzalacetone • Benzaldehyde Benzenesulfonic acid  $\blacksquare$ • Benzenesulfonyl chloride • Benzine, gasoline Benzoic acid Benzol Benzoquinone Benzotriazole Benzyl alcohol • Benzyl benzoate • Benzyl bromide A 20°C/68°F: ● Benzyl chloride, anhydrous Benzyl chloride, hydrous Benzyl chloroformate Benzyl iodide Benzylphenol Bio-diesel • Bio-ethanol Bisphenol A Bitumen • • • Black liquor (sulfate) • Black liquor (sulfide) Blood Borax, aqueous solution [= sodium tetraborate decahydrate, aqueous solution] Borax, melt (= sodium tetraborate, melt) Boron trichloride • Brake fluids, glycolbased • Bromine trifluoride Bromine, liquid Butadiene Butanal oxime Butane Butanedione • • • Butanethiol • Butanol (= butyl alcohol) Butanone (= methyl ethyl ketone) • Butin 20°C/68°F: ● Butter Buttermilk 20°C/68°F: ● Butyl acetate (= acetic acid butyl ester) Butyl acrylate 

<sup>●</sup> resistant; ■ not resistant; ▲ limited resistance; – insufficient data; p.corr = pitting corrosion

		Stainless steel	
Medium	SIGRAFLEX flexible graphite	316, 316L, 1.4401, 1.4404	Pure PTFE
Butyl chloride, anhydrous	TTEXIBLE GLAPHITE	1.4404	PUIEFIFE
Butyl chloride, hydrous	•		•
Butyl phosphine	•		•
Butyl vinyl ether	•		•
Butylamine	•		•
Butylbenzene	•		•
Butylcyclohexyl chloroformate	•		•
Butylphenol	•	20°C/68°F: ●	•
Butyltoluene	•	•	•
Butyltrichlorosilane	•		•
Butyraldehyde (= butanal)	•	•	•
Butyric acid	•		•
C		·	
Calcium acetate, aqueous solution	•	20°C/68°F: ●	•
Calcium bisulfite, aqueous solution	•	20°C/68°F: ●	•
Calcium chloride	•	<b>A</b>	•
Calcium hydroxide, hydrous			
[= caustic lime = lime water = calcium oxide, aqueous]	•	20°C/68°F: ●	•
Calcium hypochlorite	•		•
Calcium nitrate	<b>A</b>	20°C/68°F: ●	•
Calcium nitrate, aqueous solution	•	<u> </u>	•
Calcium oxide	•		•
Calcium phosphate, aqueous solution	•		•
Calcium sulfate	•	20°C/68°F: ●	•
Calcium sulfide, aqueous solution	•		•
Calcium sulfite, cold saturated	•	20°C/68°F: ●	•
Carbon dioxide	•		•
Carbon disulfide	•		•
Carbon monoxide, anhydrous	•		•
Carbon tetrachloride, anhydrous			
[= tetrachlorocarbon, anhydrous]	•		•
Carbon tetrachloride, hydrous (= tetrachlorocarbon, hydrous)	•		•
Cesium melt			
Chlorhexidine	•		•
Chlorinated lime, anhydrous	•	20°C/68°F: ●	•
Chlorinated lime, hydrous	•		•
Chlorine dioxide			•
Chlorine trifluoride			
Chlorine water			•
Chlorine, hydrous			•
Chlorine, anhydrous	400 °C / 752 °F: ●	20°C/68°F: ●	•
Chloroacetic acid (= monochloroacetic acid)	•		•
Chloroacetone	•		
Chlorobenzene, anhydrous (= monochlorobenzene, anhydrous)	•		•
Chlorobenzene, hydrous (= monochlorobenzene, hydrous)	•		•
Chlorobutadiene (= chloroprene)	•		•
Chloronitrobenzene (= nitrochlorobenzene)	•	120°C/248°F: ●	
Chlorosulfonic acid, aqueous solution 10 %	•		•
Chloropylfonia agid 100 l/		20°C/68°F: ●	•
Chromia paid agreeus solution 10 %	20.00 / 60.05	[p. corr]	•
Chromic acid, aqueous solution 10 %	20°C/68°F: ●		
Chromic acid, aqueous solution > 10 %			
Chroming solutions  Chromium (III) potagoium gulfata (- ahroma alum)			
Chromium(III) potassium sulfate (= chrome alum)			
Citric acid, aqueous solution 25 %	•		•
Claphon			
Clophen	•		

<sup>●</sup> resistant; ■ not resistant; ▲ limited resistance; – insufficient data; p.corr = pitting corrosion

		Stainless steel	
M. P.	SIGRAFLEX	316, 316L, 1.4401,	D DTEE
Medium Cohalt shlarida aguagua salutian	flexible graphite	1.4404	Pure PTFE
Cobalt chloride, aqueous solution  Cocoa butter			
Cod liver			
Coffee			
Copper acetate	•	20°C/68°F:	
Copper carbonate		20°C/68°F: •	
Copper calburate  Copper chloride, aqueous solution		20 0700 1.	
Copper cyanide, aqueous solution		20°C/68°F: ●	
Copper sulfate		20°C/68°F:	
Copper(II) nitrate, aqueous solution 50 %		20 07 00 1.0	
Copper(II) nitrate, aqueous solution > 50 %			
Creosol	•	•	
Creosote	•	•	•
Cresol	•	•	•
Crude oil	•	•	•
Cryolite (= sodium fluoroaluminate)	•		•
Cumene (= isopropyl benzene)	•	•	•
Cyclohexane	•	•	•
Cyclohexanol	•	•	•
Cyclohexanone	•	•	•
D			
Decahydronaphthalene	•	•	•
Decane	•	•	•
Dextrin, aqueous solution	•	•	•
Diacetone alcohol	•	•	•
Dibenzyl ether	•	•	•
Dibutyl ether	•	•	•
Dibutyl phthalate	•	•	•
Dibutylamine	•	•	•
Dichloro isopropyl ether	•	_	•
Dichloro hexylamine	•	_	•
Dichloroacetic acid methyl ester	•	=	•
Dichloroethene, anhydrous (= dichloroethylene, anhydrous = vinylidene dichloride, anhydrous)	•	32°C/90°F: ●	•
Dichloroethene, hydrous (= dichloroethylene, hydrous = vinylidene dichloride, hydrous)			
Dichloroethane, anhydrous (= ethylene chloride, anhydrous			
= ethylene dichloride, anhydrous]	•	20°C/68°F: ●	•
Dichloroethane, hydrous (= ethylene chloride, hydrous = ethylene dichloride, hydrous)	•		•
Dichloromethane, anhydrous [= methylene chloride, anhydrous]	•	•	•
Dichloromethane, hydrous (= methylene chloride, hydrous)	•		•
Diesel oil, Diesel fuel	•	•	•
Diethyl ketone (= 3-pentanone)	•	•	•
Diethyl ether	•	•	•
Diethyl sebacate	•	•	•
Diethyl succinate	•	•	•
Diethylene glycol	•	•	•
			▲, media causes swelling
Difluoromethane	<u> </u>		of PTFE.
Diglycolic acid, aqueous solution	•		•
Dihydroxybenzene (= (p-)hydroquinone)	•		
Diisobutyl ketone	<u> </u>		
Diisobutylene (= iso-octene)	•	•	
Diisopropyl ketone	•	•	•

lacktriangle resistant; lacktriangle not resistant; lacktriangle limited resistance; – insufficient data; p.corr = pitting corrosion

		Stainless steel	
Medium	SIGRAFLEX flexible graphite	316, 316L, 1.4401, 1.4404	Pure PTFE
Dimethyl phthalate	Ttexiste graphite	1.7707	T GICT IT E
Dimethyl sulfate (DMS)	•		•
Dimethyl urea	•	_	•
Dimethylacetamide (DMAC)	•	_	•
Dimethylamine	•	•	•
Dimethylamino sulfochloride	•	_	•
Dimethylaniline	•	•	•
Dimethylformamide (DMF)	•	•	•
Dinitrogen monoxide (= laughing gas)	20°C/68°F: ●	•	•
Dioctyl phthalate (DOP)	•	•	•
Dioctyl sebacate	•	•	•
Dioxane	•	•	•
Dipentene (= limonene)	•	•	•
Diphenyl ether	•	•	•
Dipropyl ketone	•	•	•
Disulfur dichloride, anhydrous	•	20°C/68°F: ●	•
Disulfur dichloride, hydrous	•		•
Dithiophosphoric acid	•	<u> </u>	•
Dodecyl alcohol	•	•	•
<u>E</u>			
Epichlorohydrine	•	20°C/68°F: ●	•
Ethandiole (= monoethylene glycol = ethylene glycol = glycol)	<u> </u>	<u> </u>	•
Ethane	<u> </u>	<u> </u>	•
Ethanol (= ethyl alcohol)	<u> </u>	<u> </u>	•
Ethanolamine			
[= monoethanolamine = aminoethylethanolamine]	<u> </u>		•
Ethyl acetate (= acetic acid ethyl ester)			•
Ethyl acrylate		<u> </u>	•
Ethyl benzene	<u> </u>		•
Ethyl chloride, anhydrous	•		•
Ethyl chloride, hydrous			
Ethyl formate			•
Ethyl mercaptan	•		•
Ethyl silicate			
Ethylene (= ethene)			
Ethylene chlorohydrin (= 2-chloroethanol)			
Ethylene dibromide (= 1,2-dibromoethane)			
Ethylene oxide [E0]	•, only use products with BAM test		
Ethylenediamine (EDA)	report for E0.		
Ethylhexanol	•		
F			
Fat [mineral, vegetable, animal]	•		•
Fatty acid methyl ester			
Fatty acids			
Ferric chloride, ferric(III) chloride			
Ferric(III) nitrate		20°C/68°F: ●	•
Ferric(III) sulfate, aqueous solution 10 %		20°C/68°F: ●	•
Ferric(III) sulfate, aqueous solution > 10 %	•		•
Ferric(II) sulfate, aqueous solution	•	20°C/68°F: ●	•
Ferric oxide (= iron(III) oxide)	•		•
Ferric phosphate	•	•	•
Firewater	•	•	•
Fluorine dioxide			
Fluorine, gaseous			
Fluorine, liquid			
Fluorobenzene	•	•	•

<sup>●</sup> resistant; ■ not resistant; ▲ limited resistance; – insufficient data; p.corr = pitting corrosion

Platerial suitability for specific media			
	SIGRAFLEX	Stainless steel 316, 316L, 1.4401,	
Medium	flexible graphite	1.4404	Pure PTFE
Formaldehyde (= methanal = methyl aldehyde)	•	•	•
Formalin	•	•	•
Formamide	•	•	•
Formic acid	•	20°C/68°F: ●	•
Fruit juices and fruit acids	•	•	•
Furan (= furfuran)	•	_	•
Furfuryl alcohol	•	•	•
<u>G</u>			
Gallic acid, aqueous solution			
[= trihydroxybenzoic acid, aqueous solution]	•	20°C/68°F: ●	•
Gelatin, aqueous solution	•	•	•
Glucose (= dextrose)	<u> </u>	•	•
Glycerol (= propanetriol)	<u> </u>	<u> </u>	•
Glycerol triacetate (= triacetin)			•
Glycolic acid			•
Grain alcohol	•	<u> </u>	•
Green liquor	•		•
<u>H</u>			
Heat carrier oil		•	•
Helium		•	•
Heptane	•	•	•
Heptanone		•	•
Hexachlorobutadiene, anhydrous	•	<u> </u>	•
Hexachlorobutadiene, hydrous			•
Hexachlorocyclohexane [= lindane]	•	<u> </u>	•
Hexafluorosilicic acid, not containing HF		0000/0005	
[= fluorosilicic acid, not containing HF]	<u> </u>	20°C/68°F: ●	•
Hexafluorosilicic acid, containing HF			
[= fluorosilicic acid, containing HF]		<u> </u>	•
Hexamethylenediamine (HMDA)	•	<del>-</del> -	
Hexamethylenetetramine (= urotropine)			
Hexane Hexanetriol			
Hexyl alcohol			
Hydraulic oil			
Hydrazine			
Hydrazine hydrate			
Hydrazine sulfate, aqueous solution 10 %			
Hydrazine sulfate, aqueous solution > 10 %			
Hydrazone ester	•		•
Hydrazone	•		•
Hydrazide			•
Hydrocarbons	•	•	•
Hydrochloric acid, aqueous solution	•		•
Hydrochloric acid, gaseous (= hydrogen chloride, gaseous)	•		•
Hydrochlorofluorocarbons (CFC), anhydrous	•	20°C/68°F: ●	•
Hydrochlorofluorocarbons (CFC), hydrous	•		•
Hydrocyanic acid	•	20°C/68°F: ●	•
Hydrofluoric acid, anhydrous	•		•
Hydrofluoric acid, aqueous solution 40 %			
or aqueous solution 60 %	•		•
Hydrofluorocarbons, anhydrous	•	•	•
Hydrofluorocarbons, hydrous	•		•
Hydrofluorosilicic acid	•		•
Hydrogen	•	-	•
Hydrogen bromide	•		•

lacktriangle resistant; lacktriangle not resistant; lacktriangle limited resistance; – insufficient data; p.corr = pitting corrosion

Stainless steel **SIGRAFLEX** 316, 316L, 1.4401, flexible graphite Medium 1.4404 Pure PTFE •, only use high-purity graphite in Hydrogen peroxide Z-quality, as impurities reduce the content of hydrogen peroxide. 80°C/176°F: ● [= hydrogen superoxide] 400 °C / 752 °F: ● Hydrogen sulfide, aqueous solution 4% Hydrogen sulfide, aqueous solution > 4% Hydroquinone (= p-dihydroxybenzene) Hydroxylamine sulfate, aqueous solution 10 % • Hydroxylamine sulfate, aqueous solution > 10 % lodine, anhydrous 20°C/68°F: ● 20°C/68°F: ● lodine, hydrous (p. corr) lodine tincture Isobutyl alcohol Isododecane • Isooctane Isophorone diamine Isophorone diisocyanate (IPDI) • • Isoprene Isopropanol (= isopropyl alcohol) Isopropyl acetate Isopropyl ether K Kerosene Ketone Lactam Lactic acid  $\blacksquare$ Lauryl alcohol Lead acetate 20°C/68°F: ● Lead arsenate 20°C/68°F: ● Lead nitrate, aqueous solution 20°C/68°F: ● • 20°C/68°F: ● • Lithium bromide Lithium chloride, aqueous solution Lithium melt Lysol Magnesium carbonate 20°C/68°F: ● 20°C/68°F: ● Magnesium chloride, aqueous solution 30 % (p. corr) Magnesium chloride, aqueous solution > 30 % • Magnesium hydroxide • 20°C/68°F: ● • Magnesium sulfate (= epsom salt) 20°C/68°F: ● • Maleic acid Maleic acid anhydride (MSA) 100°C/212°F: ● Malic acid, aqueous solution  $50\,\%$ Malic acid, aqueous solution > 50%Mercury 50 °C / 122 °F: ● • Mercury chloride, aqueous solution • 20°C/68°F: ● Mercury cyanide • Methacrylic acid methyl ester • Methane • • Methanol (= methyl alcohol) • Methyl acetate Methyl bromide Methyl butyl ketone

<sup>●</sup> resistant; ■ not resistant; ▲ limited resistance; – insufficient data; p.corr = pitting corrosion

Platerial suitability for specific media			
	SIGRAFLEX	Stainless steel 316, 316L, 1.4401,	
Medium	flexible graphite	1.4404	Pure PTFE
Methyl chloride, anhydrous [= chloromethane, anhydrous = monochloromethane, anhydrous]	•	•	•
Methyl chloride, hydrous (= chloromethane,			
hydrous = monochloromethane, hydrous)	<u> </u>		•
Methyl ether	<u> </u>		•
Methyl formate			•
Methyl glycol acetate			•
Methyl isobutyl ketone			•
Methyl isopropyl ketone			•
Methyl methacrylate			•
Methyl oleate			•
Methyl sulfuric acid, aqueous solution			•
Methylamine, aqueous solution		· <del>-</del> -	•
Methylcyclohexanol			•
Methylcyclopentane			•
Milk		70°C/158°F: ●	•
Mine gas			•
Mineral oil			•
Mineral water			•
Molasses			•
Mono bromobenzene, anhydrous			•
Mono bromobenzene, hydrous			•
Morpholine			
N			
Naphthalene		·	•
Naphtha			•
Natural gas			•
Neon		00000000	•
Nickel acetate, aqueous solution		20°C/68°F: ●	•
Nickel chloride		0000/0005	•
Nickel nitrate, aqueous solution		20°C/68°F: ●	•
Nickel sulfate, aqueous solution	0000 (0005	100 °C / 212 °F: ●	•
Nitric acid, aqueous solution 10 %		100 °C / 212 °F: ●	•
Nitric acid, aqueous solution 37%		100 °C / 212 °F: ●	•
Nitric acid, aqueous solution 65 %		20°C/68°F: ●	
Nitric acid fuming			
Nitro thinner			•
Nitrobenzene			•
Nitroethane			
Nitrogen			
Nitrogen oxides, hydrous [= nitrous fumes, hydrous = NOx, hydrous]			•
Nitrogen oxides, anhydrous	400 C/752°F: ●,		
[= nitrous fumes, anhydrous = NOx, anhydrous]	but please always ask for advice.		•
Nitrogen tetroxide			•
Nitrohydrochloric acid			•
Nitromethane	<u> </u>		•
Nitropropane	<u> </u>		•
Nitrosulfuric acid			•
Nitrotoluene		•	•
Nonane	<u> </u>	<u> </u>	•
Norbornadiene (= bicycloheptadiene)	<u> </u>	•	•
0			
Octadiene		•	•
Octane			•
Octanol (= octane alcohol)	<u> </u>		•
Octyl cresol		•	•

<sup>●</sup> resistant; ■ not resistant; ▲ limited resistance; – insufficient data; p.corr = pitting corrosion

		SIGRAFLEX	Stainless steel 316, 316L, 1.4401,	
Olice acid   300.°C/572°F;   Olicem   sulfur trioxide dissolved in sulfuric acid	Medium			Pure PTF
Dieum   F. suffur trivaxide dissolved in sulfuric solid	Oil (mineral, vegetable, animal)		•	
### ### #############################	Oleic acid	<u> </u>	300°C/572°F: ●	•
Oxelic acid, aqueous solution 10 %         Q 0°C / 68 °F; €           Oxalic acid, aqueous solution > 10 %         20°C / 482 °F; €           Company of temperature resistance and BM test report needs to be considered.         0           Oxygen         100°C / 212 °F; €           In higher temperatures please refer to our technical information about temperatures please refer to our technical information about temperature special resistance and ask for advice.         0           Paraffin         €         0           Paraffin         €         0           Paraffin         €         0           Pentateulorophanol         €         0           Pentateulorophanol         €         0           Pentatene         €         0           Pen	Oleum (= sulfur trioxide dissolved in sulfuric acid)			•
Disable acid, aqueous solution > 10%	Orthoboric acid	•	•	•
SQL technical information about temperature resistance and SM test raport needs to be considered.	Oxalic acid, aqueous solution 10 %	<u> </u>	20°C/68°F: ●	•
SSL technical information about temperature resistance and BM test report needs to be considered.	Oxalic acid, aqueous solution > 10 %	•		•
December   December	Oxygen	SGL technical information about temperature resistance and BAM test report needs to be considered.	•	•
Palmitic acid	Ozono or ozono pir miyturo	for higher temperatures please refer to our technical information about tempe-		
Parintic acid				
Partafiln	•			_
Pentachlorophenot         ●				
Pentafluoroethane         ● ○ ○ ○ ○ ○ ○ ○ ○ ○ ○ ○ ○ ○ ○ ○ ○ ○ ○ ○				
Pentane         ● (a)	·			
Pertolaric acid				
Perchloric acid				•
Pertolaum         Image: Company of the company	Pentene			•
Petroleum Her Petroleum ether	Davablasia asid		_	
Petroleum ether         ●           Phenol (= carbolic acid)         ●           Phenyl benzene         ●           Phenyl benzene         ●           Phenylhydrazine         ●           Phosgene, anlydrous         ●           Phospene, hydrous         ●           Phosphoric acid, aqueous solution >20%         ●           Phosphoric acid, aqueous solution 20%         ●           Phosphoric acid, impure         ●           Phosphoric acid, impure         ●           Phosphorus daychloride         ●           Phosphorus trichloride         ●           Phosphorus trichloride         ●           Phthalic acid         ●           Phthalic acid anhydride (PSA)         ●           Plicric acid, aqueous solution         ●           Ployethersulfone         ●           Polyethersulfone         ●           Potassium acetate         ●           Potassium bromate, aqueous solution         ●           Potassium bromate, aqueous solution         ●           Potassium bromate, aqueous solution         ●           Potassium carbonate (= potash)         ●           Potassium carbonate (= potash)         ●           Potassium chloride				•
Phenol [= carbolic acid]         ●           Phenyl benzene         ●           Phenylethyl ether         ●           Phenylethyl ether         ●           Phospene, anhydrous         20°C/68°F; ●           Phospene, hydrous         ●           Phosphoric acid, aqueous solution >20%         ●           Phosphoric acid, aqueous solution 20%         ●           Phosphoric acid, impure         ●           Phosphoric acid, impure         ●           Phosphoric acid, impure         ●           Phosphory acychloride         ●           Phosphory acychloride         ●           Phosphory acychloride         ●           Phthalic acid         20°C/68°F; ●           Phthalic acid anhydride (PSA)         ●           Phthalic acid anhydride (PSA)         ●           Plicric acid, aqueous solution         ●           Potassium acetate         ●           Potassium bromide, aqueous solution         ●           Potassium bromate, aqueous solution         ●           Potassium bromide, aqueous solution         ●           Potassium carbonate (= potash)         ●           Potassium chloride         ●           Potassium chloride         ●				•
Phenyl benzene         ●           Phenylethyl ether         ●           Phenylhydrazine         ●           Phosgene, anhydrous         ●           Phosgene, hydrous         ●           Phosphoric acid, aqueous solution > 20%         ●           Phosphoric acid, aqueous solution 20%         ●           Phosphoric acid, impure         ●           Phosphoric acid, impure         ●           Phosphorus trichloride         ●           Phosphorus trichloride         ●           Phthalic acid         ●           Phthalic acid         ●           Phthalic acid, aqueous solution         ●           Picric acid, aqueous solution         ●           Piperidine         ●           Polyethersulfone         ●           Potassium acetate         ●           Potassium bifluoride, saturated         ●           Potassium bromate, aqueous solution         ●           Potassium bromate, aqueous solution         ●           Potassium bromate (= potash)         ●           Potassium carbonate (= potash)         ●           Potassium chloride         ●           Potassium chloride         ●           Potassium cyanate         ●				•
Phenylethyl ether         ●           Phenylhydrazine         ●           Phosgene, anhydrous         ●           Phosgene, hydrous         ●           Phosphoric acid, aqueous solution > 20%         ●           Phosphoric acid, aqueous solution 20%         ●           Phosphoric acid, impure         ●           Phosphoric acytholoride         ●           Phosphorus dxychloride         ●           Phosphorus trichloride         ●           Phthalic acid         ●           Phthalic acid anhydride (PSA)         ●           Picric acid, aqueous solution         ●           Piperidine         ●           Polyethersulfone         ●           Potassium acetate         ●           Potassium bifluoride, saturated         ●           Potassium biromate, aqueous solution         ●           Potassium bromate, aqueous solution         ●           Potassium bromide, aqueous solution         ●           Potassium carbonate (= potash)         ●           Potassium chlorate, aqueous solution         ●           Potassium chlorate, aqueous solution         ●           Potassium chlorate, aqueous solution         ●           Potassium chlorate         ●				•
Phenylhydrazine         ● 20 °C / 68 °F: ●           Phosgene, anhydrous         ● 20 °C / 68 °F: ●           Phosphoric acid, aqueous solution > 20%         ● 20 °C / 68 °F: ●           Phosphoric acid, aqueous solution 20%         ● 100 °C / 212 °F: ●           Phosphoric acid, impure         ● A           Phosphorus oxychloride         ● Inceptor of the control o	•			•
Phosgene, hydrous         ■           Phospene, hydrous         ■           Phosphoric acid, aqueous solution >20%         ● 20°C/68°F; ●           Phosphoric acid, aqueous solution 20%         ● 100°C/212°F; ●           Phosphoric acid, impure         ●           Phosphorus oxychloride         ●           Phosphorus trichloride         ●           Phthalic acid         ● 20°C/68°F; ●           Phthalic acid anhydride (PSA)         ● 20°C/68°F; ●           Picric acid, aqueous solution         ● 20°C/68°F; ●           Piperidine         ● 20°C/68°F; ●           Potassium acetate         ● 20°C/68°F; ●           Potassium bifluoride, saturated         ● 20°C/68°F; ●           Potassium bromate, aqueous solution         ■ 20°C/68°F; ●           Potassium bromate, aqueous solution         ■ 20°C/68°F; ●           Potassium carbonate [= potash]         ■ 100°C/212°F; ●           Potassium carbonate [= potash]         ■ 20°C/68°F; ●           Potassium chloride         ● [p. corr]           Potassium chloride         ● [p. corr]           Potassium chromate         ● 20°C/68°F; ●           Potassium cyanate         ● 20°C/68°F; ●           Potassium cyanide [= cyan potassium]         ● 20°C/68°F; ●           Potassium hexacyanoferrate II				•
Phosgene, hydrous         20 °C / 68 °F: ○           Phosphoric acid, aqueous solution > 20 %         20 °C / 68 °F: ○           Phosphoric acid, aqueous solution 20 %         100 °C / 212 °F: ○           Phosphoric acid, impure         ○           Phosphorus oxychloride         ○           Phosphorus trichloride         ○           Phthalic acid         ○           Phthalic acid anhydride (PSA)         ○           Plicric acid, aqueous solution         ○           Picric acid, aqueous solution         ○           Ployethersulfone         ○           Potassium acetate         ○           Potassium bifluoride, saturated         ○           Potassium bromate, aqueous solution         ○           Potassium bromate, aqueous solution         ○           Potassium bromate, aqueous solution         ○           Potassium carbonate (= potash)         ○           Potassium carbonate (= potash)         ○           Potassium chlorate, aqueous solution         ○           Potassium chlorate         ○           Potassium chlorate         ○           Potassium chlorate         ○           Potassium chromate         ○           Potassium cyanate         ○           Potassium cyan				•
Phosphoric acid, aqueous solution > 20%         20 °C /68 °F: ●           Phosphoric acid, aqueous solution 20%         100 °C / 212 °F: ●           Phosphoric acid, impure         A           Phosphorus oxychloride         ●           Phosphorus trichloride         ●           Phthalic acid         20 °C /68 °F: ●           Phthalic acid anhydride [PSA]         ●           Picric acid, aqueous solution         20 °C /68 °F: ●           Piperidine         ●           Polyethersulfone         ●           Potassium acetate         20 °C /68 °F: ●           Potassium bifluoride, saturated         ●           Potassium bromate, aqueous solution         ●           Potassium bromate, aqueous solution         ●           Potassium bromide, aqueous solution         ●           Potassium carbonate [= potash]         ●           Potassium chlorate, aqueous solution         ② °C /68 °F: ●           Potassium chloride         ●           Potassium chlorate         ●           Potassium chlorate         ●           Potassium chromate         ●           Potassium cyanate         ●           Potassium cyanate         ●           Potassium cyanate         ●           Potassium				•
Phosphoric acid, aqueous solution 20%         100 °C/212 °F: ●           Phosphoric acid, impure         A           Phosphorus oxychloride         ■           Phosphorus trichloride         ■           Phthalic acid         20 °C/68 °F: ●           Phthalic acid anhydride (PSA)         ■ -           Picric acid, aqueous solution         20 °C/68 °F: ●           Piperidine         ■           Potassium acetate         ■           Potassium bifluoride, saturated         ■           Potassium bifluoride, saturated         ■           Potassium bromate, aqueous solution         ■           Potassium bromate, aqueous solution         ■           Potassium bromide, aqueous solution         ■           Potassium carbonate [= potash]         100 °C/212 °F: ●           Potassium carbonate [= potash]         100 °C/212 °F: ●           Potassium chloride         100 °C/212 °F: ●           Potassium chloride         (p. corr)           Potassium chromate         ■           Potassium cyanate         ●           Potassium cyan	·			•
Phosphoric acid, impure         A           Phosphorus oxychloride         ■           Phosphorus trichloride         ■           Phthalic acid         20 °C /68 °F; ●           Phthalic acid anhydride [PSA]         ■           Picric acid, aqueous solution         20 °C /68 °F; ●           Piperidine         ■           Polyschersulfone         ■           Potassium bifluoride, saturated         ■           Potassium bifluoride, saturated         ■           Potassium bromate, aqueous solution         ■           Potassium bromate, aqueous solution         ■           Potassium carbonate [= potash]         100 °C /212 °F; ●           Potassium carbonate [= potash]         20 °C /68 °F; ●           Potassium chlorate, aqueous solution         20 °C /68 °F; ●           Potassium chlorate, aqueous solution         20 °C /68 °F; ●           Potassium chlorate [= potash]         100 °C /212 °F; ●           Potassium chlorate         [p. corr]           Potassium chromate         ■           Potassium cyanate         ■           Potassium cyanate [= cyan potassium]         20 °C /68 °F; ●           Potassium formate         ■           Potassium hexacyanoferrate III         ■				•
Phosphorus oxychloride         ■           Phosphorus trichloride         ■           Phthalic acid         20°C/68°F: ●           Phthalic acid anhydride [PSA]         ■           Picric acid, aqueous solution         ●           Piperidine         ●           Polyethersulfone         ●           Potassium acetate         ●           Potassium bifluoride, saturated         ●           Potassium bromate, aqueous solution         ●           Potassium bromate, aqueous solution         ●           Potassium bromide, aqueous solution         ●           Potassium carbonate [= potash]         ●           Potassium chlorate, aqueous solution         ●           Potassium chlorate, aqueous solution <td></td> <td><u>_</u></td> <td></td> <td>•</td>		<u>_</u>		•
Phosphorus trichloride         ■           Phthalic acid         20 °C /68 °F · ●           Phthalic acid anhydride [PSA]         □           Picric acid, aqueous solution         20 °C /68 °F · ●           Piperidine         □           Potypethersulfone         □           Potassium acetate         □           Potassium bifluoride, saturated         □           Potassium borate, aqueous solution         □           Potassium bromate, aqueous solution         □           Potassium bromide, aqueous solution         □           Potassium carbonate [= potash]         □           Potassium chlorate, aqueous solution         □           □         □           Potassium chlorate, aqueous solution         □           □         □           □         □           □         □           □         □           <				•
Phthalic acid         20 °C / 68 °F: ●           Phthalic acid anhydride (PSA)         -           Picric acid, aqueous solution         20 °C / 68 °F: ●           Piperidine         ●           Potypethersulfone         ●           Potassium acetate         ●           Potassium bifluoride, saturated         ●           Potassium borate, aqueous solution         ●           Potassium bromate, aqueous solution         ●           Potassium bromide, aqueous solution         ●           Potassium carbonate (= potash)         ●           Potassium chlorate, aqueous solution         20 °C / 68 °F: ●           Potassium chlorate, aqueous solution         20 °C / 68 °F: ●           Potassium chlorate         ●           Potassium chloride         ●           Potassium chromate         ●           Potassium cyanate         ●           Potassium cyanide (= cyan potassium)         ●           Potassium formate         ●           Potassium hexacyanoferrate III         ●	. ,	<u> </u>		•
Phthalic acid anhydride (PSA)         □ <td< td=""><td>Phosphorus trichloride</td><td><u> </u></td><td></td><td>•</td></td<>	Phosphorus trichloride	<u> </u>		•
Picric acid, aqueous solution         20 °C / 68 °F: ●           Piperidine         ●           Polyethersulfone         ●           Potassium acetate         20 °C / 68 °F: ●           Potassium bifluoride, saturated         ●           Potassium borate, aqueous solution         ●           Potassium bromate, aqueous solution         ■           Potassium bromide, aqueous solution         [p. corr]           Potassium carbonate (= potash)         100 °C / 212 °F: ●           Potassium chlorate, aqueous solution         20 °C / 68 °F: ●           Potassium chloride         (p. corr]           Potassium chloride         ●           Potassium cyanate         ●           Potassium cyanide (= cyan potassium)         ●           Potassium formate         ●           Potassium formate         ●           Potassium hexacyanoferrate III         ●			20°C/68°F: ●	•
Piperidine         ●         −           Polyethersulfone         ●         −           Potassium acetate         ●         20 °C / 68 °F: ●           Potassium bifluoride, saturated         ●         −           Potassium bornate, aqueous solution         ●         −           Potassium bromate, aqueous solution         ●         (p. corr)           Potassium carbonate [= potash]         ●         100 °C / 212 °F: ●           Potassium chlorate, aqueous solution         20 °C / 68 °F: ●         20 °C / 68 °F: ●           Potassium chloride         ●         (p. corr)           Potassium chromate         ●         ●           Potassium cyanate         ●         20 °C / 68 °F: ●           Potassium cyanide [= cyan potassium]         ●         20 °C / 68 °F: ●           Potassium formate         ●         0 °C / 68 °F: ●           Potassium hexacyanoferrate III         ●         ●				•
Polyethersulfone         -           Potassium acetate         20 °C / 68 °F: ●           Potassium bifluoride, saturated         ■           Potassium borate, aqueous solution         -           Potassium bromate, aqueous solution         20 °C / 68 °F: ●           Potassium bromide, aqueous solution         [p. corr]           Potassium carbonate (= potash)         100 °C / 212 °F: ●           Potassium chlorate, aqueous solution         20 °C / 68 °F: ●           Potassium chloride         [p. corr]           Potassium chromate         ■           Potassium cyanate         20 °C / 68 °F: ●           Potassium cyanide (= cyan potassium)         20 °C / 68 °F: ●           Potassium formate         20 °C / 68 °F: ●           Potassium hexacyanoferrate III         ●	•	<u> </u>	20°C/68°F: ●	•
Potassium acetate         20 °C / 68 °F: ●           Potassium bifluoride, saturated         ■           Potassium borate, aqueous solution         -           Potassium bromate, aqueous solution         20 °C / 68 °F: ●           Potassium bromide, aqueous solution         [p. corr]           Potassium carbonate [= potash]         100 °C / 212 °F: ●           Potassium chlorate, aqueous solution         20 °C / 68 °F: ●           Potassium chloride         [p. corr]           Potassium chromate         ■           Potassium cyanate         20 °C / 68 °F: ●           Potassium cyanide [= cyan potassium]         20 °C / 68 °F: ●           Potassium formate         20 °C / 68 °F: ●           Potassium hexacyanoferrate III         ●	•	<u> </u>		•
Potassium bifluoride, saturated       ■         Potassium borate, aqueous solution       −         Potassium bromate, aqueous solution       20 °C / 68 °F: ●         Potassium bromide, aqueous solution       [p. corr]         Potassium carbonate [= potash]       100 °C / 212 °F: ●         Potassium chlorate, aqueous solution       20 °C / 68 °F: ●         Potassium chloride       [p. corr]         Potassium chromate       ■         Potassium cyanate       20 °C / 68 °F: ●         Potassium cyanide [= cyan potassium]       20 °C / 68 °F: ●         Potassium formate       20 °C / 68 °F: ●         Potassium hexacyanoferrate III       ●	,			•
Potassium borate, aqueous solution  Potassium bromate, aqueous solution  Potassium bromide, aqueous solution  Potassium bromide, aqueous solution  Potassium carbonate [= potash]  Potassium chlorate, aqueous solution  Potassium chlorate, aqueous solution  Potassium chlorate  Potassium chloride  Potassium chromate  Potassium cyanate  Potassium cyanate  Potassium cyanide [= cyan potassium]  Potassium formate  Potassium formate  Potassium formate  Potassium hexacyanoferrate III			20°C/68°F: ●	•
Potassium bromate, aqueous solution  Potassium bromide, aqueous solution  Potassium carbonate (= potash)  Potassium chlorate, aqueous solution  Potassium chlorate, aqueous solution  20 °C / 68 °F: ●  20 °C / 68 °F: ●  20 °C / 68 °F: ●  Potassium chromate  Potassium cyanate  Potassium cyanate  Potassium cyanide (= cyan potassium)  Potassium formate  Potassium formate  Potassium hexacyanoferrate III	·			•
Potassium bromide, aqueous solution  Potassium carbonate [= potash]  Potassium chlorate, aqueous solution  Potassium chlorate, aqueous solution  20 °C / 68 °F: ●  20 °C / 68 °F: ●  20 °C / 68 °F: ●  Potassium chloride  Potassium chromate  Potassium cyanate  Potassium cyanate  Potassium cyanide [= cyan potassium)  Potassium formate  Potassium formate  Potassium hexacyanoferrate III	· •	<u> </u>		•
Potassium bromide, aqueous solution       [p. corr]         Potassium carbonate [= potash]       100 °C/212 °F: ●         Potassium chlorate, aqueous solution       20 °C/68 °F: ●         Potassium chloride       [p. corr]         Potassium chromate       ■         Potassium cyanate       20 °C/68 °F: ●         Potassium cyanide [= cyan potassium]       20 °C/68 °F: ●         Potassium formate       20 °C/68 °F: ●         Potassium hexacyanoferrate III       ●	Potassium bromate, aqueous solution		<u> </u>	•
Potassium carbonate (= potash)         100 °C/212 °F: ●           Potassium chlorate, aqueous solution         20 °C/68 °F: ●           20 °C/68 °F: ●         20 °C/68 °F: ●           Potassium chloride         (p. corr)           Potassium chromate         ■           Potassium cyanate         20 °C/68 °F: ●           Potassium cyanide (= cyan potassium)         20 °C/68 °F: ●           Potassium formate         20 °C/68 °F: ●           Potassium hexacyanoferrate III         ●			20°C/68°F: ●	
Potassium chlorate, aqueous solution         20 °C / 68 °F: ●         20 °C / 68 °F: ●           Potassium chloride         [p. corr]           Potassium chromate         ■         ■           Potassium cyanate         20 °C / 68 °F: ●         ●           Potassium cyanide (= cyan potassium)         20 °C / 68 °F: ●         ●           Potassium formate         20 °C / 68 °F: ●         ●           Potassium hexacyanoferrate III         ●         ●		<u> </u>		•
Potassium chloride				•
Potassium chloride       (p. corr)         Potassium chromate       ■         Potassium cyanate       ■       20 °C / 68 °F: ●         Potassium cyanide (= cyan potassium)       ■       20 °C / 68 °F: ●         Potassium formate       ■       20 °C / 68 °F: ●         Potassium hexacyanoferrate III       ■       ●	Potassium chlorate, aqueous solution	20°C/68°F: ●	20°C/68°F: ●	•
Potassium chromate       ▲       ■         Potassium cyanate       • 20 °C / 68 °F: •         Potassium cyanide (= cyan potassium)       • 20 °C / 68 °F: •         Potassium formate       • 20 °C / 68 °F: •         Potassium hexacyanoferrate III       • •			20°C/68°F: ●	
Potassium cyanate       20 °C / 68 °F: ●         Potassium cyanide (= cyan potassium)       20 °C / 68 °F: ●         Potassium formate       20 °C / 68 °F: ●         Potassium hexacyanoferrate III       ●	Potassium chloride	<u> </u>	(p. corr)	•
Potassium cyanide [= cyan potassium]       20 °C / 68 °F: ●         Potassium formate       20 °C / 68 °F: ●         Potassium hexacyanoferrate III       ●	Potassium chromate			•
Potassium formate	Potassium cyanate	•	20°C/68°F: ●	•
Potassium hexacyanoferrate III • •	Potassium cyanide (= cyan potassium)		20°C/68°F: ●	•
	Potassium formate	•	20°C/68°F: ●	•
Potassium hydroxide, solid	Potassium hexacyanoferrate III	•	•	•
<u> </u>	Potassium hydroxide, solid	<u> </u>	•	•

<sup>●</sup> resistant; ■ not resistant; ▲ limited resistance; – insufficient data; p.corr = pitting corrosion

Place hat suitability for specific media			
	SIGRAFLEX	Stainless steel	
Medium	flexible graphite	316, 316L, 1.4401, 1.4404	Pure PTFE
Potassium hydroxide, aqueous solution	•	100°C/212°F: ●	•
Potassium hydroxide, melt	_		
Potassium hypochloride	•		•
Potassium hypochlorite	•		•
		20°C/68°F: ●	
Potassium iodide	•	[p. corr]	•
Potassium melt	350°C/662°F: ●		
Potassium nitrate, aqueous solution	•	20°C/68°F: ●	•
Potassium nitrate, melt [= saltpeter, melt]		340 °C / 644 °F: ●	
Potassium oxalate		20°C/68°F: ●	•
Potassium permanganate	20°C/68°F: ●	100 °C / 212 °F: ●	•
Potassium silicate (= water glass)		100 °C/212 °F: ●	•
Potassium sulfate, aqueous solution	•	20°C/68°F: ●	•
Printer's acetate		100 °C / 212 °F: ●	•
Propane		100 07212 1.0	•
Propanol (= propyl alcohol)			•
Propene (= propylene)			•
Propionic acid			
Propyl acetate			•
			•
Propylamine Propylamine			•
Propylene glycol			
Describer a suida (DO)	•, only use products with BAM		
Propylene oxide [P0]	test report for PO.		
Pyridine			
Pyrrole			•
S Colorador and a colorador a colorador and a colorador and a colorador and a colorador and a			
Sal volatile, cold saturated			•
Salicylic acid			•
		20°C/68°F: ●	
Salt water (= seawater)		[p. corr]	
Silicone greases, silicone oils			•
Silver nitrate, aqueous solution 10 %		100°C/212°F: ●	•
Silver nitrate, aqueous solution > 10 %	•		•
Silver nitrate, melt	350°C/662°F: ●	250 °C / 482 °F: ●	
Soap			•
Sodium acetate		20°C/68°F: ●	•
Sodium aluminate			•
Sodium ammonium hydrogen phosphate		100°C/212°F: ●	•
Sodium benzoate, aqueous solution		100°C/212°F: ●	•
Sodium borate, aqueous solution			•
Sodium carbonate (= soda)	•	100°C/212°F: ●	•
Sodium chlorate, aqueous solution 30 %	20°C/68°F: ●	100°C/212°F: ●	•
Sodium chlorate, aqueous solution > 30 %	20°C/68°F: ●		•
		20°C/68°F: ●	
Sodium chloride (= cooking salt)		[p. corr]	•
Sodium chlorite, aqueous solution			•
Sodium cyanide	•	100°C/212°F: ●	•
Sodium fluoride, aqueous solution 5 %	•	20°C/68°F: ●	•
Sodium fluoride, aqueous solution > 5 %	•		•
Sodium hydrogen carbonate (= sodium bicarbonate)	•	20°C/68°F: ●	•
Sodium hydrogen phosphate		100°C/212°F: ●	•
Sodium hydrogen sulfate (= sodium bisulfate)	•		•
Sodium hydrogen sulfate, aqueous solution 10 %			
[= sodium bisulfate, aqueous solution 10 %]	•	100°C/212°F: ●	•
Sodium hydrogen sulfate, aqueous solution > 10 %			
[= sodium bisulfate > 10 %]			•

<sup>●</sup> resistant; ■ not resistant; ▲ limited resistance; – insufficient data; p.corr = pitting corrosion

Stainless steel **SIGRAFLEX** 316, 316L, 1.4401, flexible graphite Medium 1.4404 Pure PTFE Sodium hydrogen sulfite, aqueous solution > 50%. [= sodium bisulfite, aqueous solution > 50 %] Sodium hydrogen sulfite, aqueous solution 50 % 100 °C / 212 °F: ● [= sodium bisulfite, aqueous solution 50%] 100 °C / 212 °F: ● Sodium hydrogen sulfite (= sodium bisulfite) Sodium hydroxide, aqueous solution 25 % 20°C/68°F: ● [= caustic soda 25% = caustic soda solution 25%] Sodium hydroxide, aqueous solution > 25% [= caustic soda > 25 % = caustic soda solution > 25 %] Sodium hypochloride Sodium hypochlorite Sodium methoxide Sodium melt 350 °C / 662 °F: ● Sodium nitrate, melt (= Chile saltpeter, melt) 310 °C / 590 °F: ● Sodium nitrate, aqueous solution [= Chile saltpeter, aqueous solution] 20°C/68°F: ● Sodium nitrite, saturated 100 °C / 212 °F: ● Sodium peroxide (= sodium superoxide) 100 °C / 212 °F: ● Sodium phosphate, dibasic 100 °C / 212 °F: ● Sodium phosphate, tribasic Sodium silicate (= water glass) 100 °C / 212 °F: ● Sodium sulfate, [e.g. Glauber's salt sodium sulfate decahydrate] 20°C/68°F: ● Sodium sulfide 100°C/212°F: ● Sodium sulfite, aqueous solution 50% Sodium sulfite, aqueous solution > 50 % Sodium thiosulfate, aqueous solution 25% 100 °C / 212 °F: ● Sodium thiosulfate, aqueous solution > 25 % Starch, aqueous solution Steam Stearic acid Styrene Succinic acid Succinic acid ester • Sugar, aqueous solution • 100°C/212°F: ● Sulfur chloride, anhydrous Sulfur chloride, hydrous Sulfur dichloride, anhydrous 100 °C / 212 °F: ● Sulfur dichloride, hydrous 100 °C / 212 °F: ● Sulfur dioxide, anhydrous Sulfur dioxide, hydrous Sulfur hexafluoride • • Sulfur trioxide Sulfur, anhydrous, liquid 130 °C / 266 °F: ● Sulfur, hydrous 20°C/68°F: ● Sulfuric acid 98 % 100 °C/212 °F: ● Sulfuric acid, aqueous solution < 70 % 100°C/212°F: ● Sulfurous acid 100°C/212°F: ● т Tannic acid (= tannin) Tetrabromomethane Tetrachloroethane, anhydrous Tetrachloroethane, hydrous Tetrachloroethylene, anhydrous (= perchloroethylene, anhydrous) Tetrachloroethylene, hydrous (= perchloroethylene, hydrous) Tetrafluoroboric acid, containing HF 

<sup>•</sup> resistant; ■ not resistant; ▲ limited resistance; - insufficient data; p.corr = pitting corrosion

		Stainless steel	
Medium	SIGRAFLEX flexible graphite	316, 316L, 1.4401, 1.4404	Pure PTFE
Tetrafluoroboric acid, not containing HF	Proxibite graphite		1 4101 112
1,1,1,2-Tetrafluoroethane	•		•
Tetrahydrofuran	•	•	
Tetralin (= 1,2,3,4-tetrahydronaphthalene)	•	•	
Thionyl chloride	•		•
Tin chloride, aqueous solution	•		
Tin, liquid			•
Titanium tetrachloride	•		
Toluene	•	•	
Tributyl phosphate	•	•	
Tricalcium phosphate	•	•	
Trichloroacetic acid	•		•
Trichloroethane, anhydrous	•	20°C/68°F: ●	•
Trichloroethane, hydrous	•	20 07 00 1.0	•
Trichloroethene, anhydrous (= trichloroethylene, anhydrous)	•	100 °C/212 °F: ●	
Trichloroethene, hydrous (= trichloroethylene, hydrous)	•	100 07 212 11 0	
Trichloromethane, anhydrous [= chloroform, anhydrous]			
Trichloromethane, hydrous [= chloroform, hydrous]			
Trichlorosilane	•		
montorositano			A
Trichlorotrifluoroethane, anhydrous			▲, media causes swelling
[= Freon 113, anhydrous]		•	of PTFE.
(=1100H110, dHHydrodo)			
Trichlorotrifluoroethane, hydrous			▲, media causes swelling
[= Freon 113, hydrous]			of PTFE.
Triethanolamine			OI FILE.
Triethylamine	•		
Triethylenetetramine (TETA)			
Trimethylaluminium	•		
trioctyl phosphate	•		
Trisodium phosphate	•		
Turpentine	•		
U			
Uranium hexafluoride			
Urea		•	
V			
Vaseline			
Vinegar, aqueous solution 50 % (= wine vinegar)		100°C/212°F: ●	
Vinyl chloride (= chloroethene)	•	100 07212 1.0	
W			
Water	•		•
Water, purified	•		
Wine		•	
X			
Xenon	•		
Xylene	•		
Y			
Yeast, aqueous solution	•		
Z			
Zinc acetate, aqueous solution	•	100°C/212°F: ●	
Zinc acetate, aqueous solution	•	100 0/212 F.	
Zinc chloride, aqueous solution Zinc cyanide	•	20°C/68°F:	
Zinc syanite Zinc sulfate, aqueous solution	•	100 °C/212 °F: ●	
Zinc surface, aqueous solution Zinc, liquid	•	100 C/ZIZ F.	
Zirio, iiyuiu	•		

lacktriangle resistant; lacktriangle not resistant; lacktriangle limited resistance; – insufficient data; p.corr = pitting corrosion



Additional information on our SIGRAFLEX sealing materials can be found under "Download Center" on our homepage.
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